

STRUCTURE AND CONFORMATION OF CIS AND TRANS-3,5-DIMETHYLVALEROLACTONES¹

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(Received in USA 6 August 1970; received in UK for publication 14 September 1970)

The study of the structure and conformation of saturated δ -lactones has recently attracted considerable attention.²⁻⁸ The relationship between the structure and conformation of these compounds to their NMR, IR, CD and ORD properties has been studied. There is general agreement that the C-O-C group is planar in δ -lactones;⁵⁻⁷ however, there is disagreement on the preferred conformation of this ring. On the basis of x-ray analysis of a bromodilactone it was implied that the presence of the C-O-C group in a 6-membered ring produced a boat conformation,⁶ and ORD data was interpreted as supporting this view.⁸ More recently, however, the IR⁷ and NMR^{2,3} spectral data of some substituted δ -lactones are more consistent with a half-chair conformation. This growing interest in the stereochemistry of δ -lactones prompted us to prepare and examine the NMR properties of cis and trans-3,5-dimethylvalerolactone (I), a diastereoisomeric pair of lactones, which has a combination of substitutes that has not previously been studied.* The NMR spectra of the two lactones in CCl₄ are shown in Figures Ia and IIa. Significant aspects of the spectra are the shifts in the two spectra of the C₃-CH₃ (δ 1.01 to 1.09) and C₅-H (δ 4.29 to 4.46) and the appearance of the C₅-CH₃ group at δ 1.33 in both spectra. These results can be explained if the NMR spectrum having the highest field resonance for the C₃-CH₃ is assigned to the cis-isomer and the cis and trans isomer are represented by half-chair conformation. The cis-isomer has the C₃-CH₃ and C₅-CH₃ groups in a cis-1,3 relationship and may be expected to exist preferentially in conformation IA which avoids diaxial opposition of these groups.** Since the cis and trans isomer have the same resonance for the C₅-CH₃ group, the trans isomer is assigned conformation IB which has this group in an equatorial position.**

*The synthesis of these lactones and their conversion to 5-alkyl-5-(3'-hydroxy-1'-methylbutyl)-barbituric acids will be reported elsewhere.

**Only one of the optical isomers is shown.

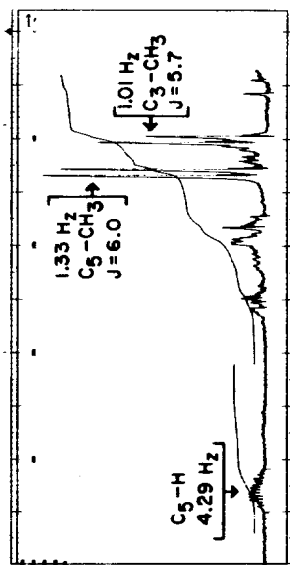


Figure 1a - 100 MHz spectrum of 1A

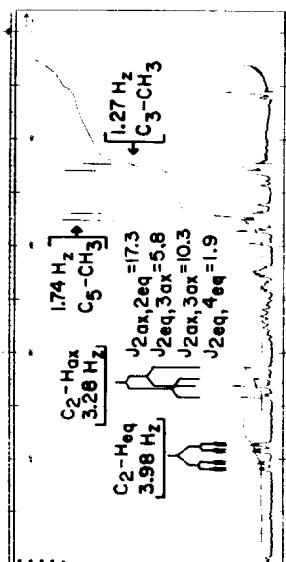
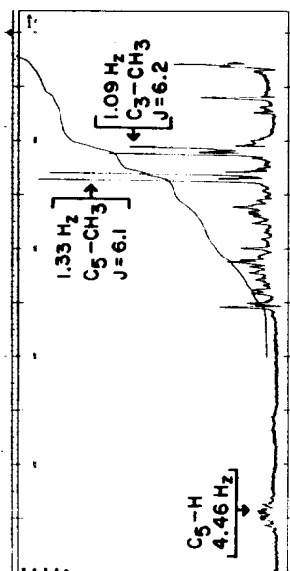
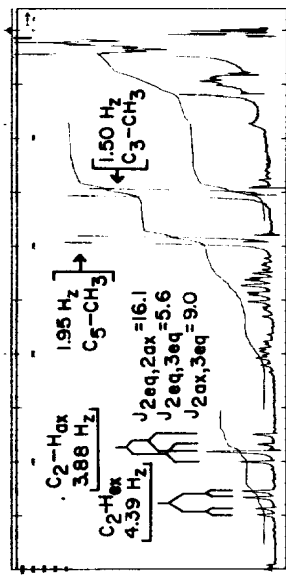
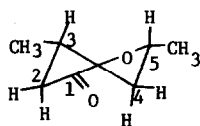
Figure 1b - 100 MHz spectrum of 1A with 0.23 mole % of $Eu(DPM)_3$ added

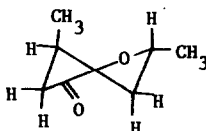
Figure 1Ia - 100 MHz spectrum of 1B

Figure 1Ib - 100 MHz spectrum of 1B with 0.24 mole % of $Eu(DPM)_3$ added

The shift of δ 1.01 to 1.09 and the increase in J_{H,CH_3} for the C_3-CH_3 group is the results expected on going from an equatorial to an axial CH_3 -group,^{9,*} and the shift of 4.29 to 4.46 Hz of the C_5-H can be explained by the cis 1,3-diaxial interaction of the C_3-CH_3 , C_5-H of IB which is absent in IA.^{10,**} Proof for the correctness of these assignments was sought by observing the NMR spectra of IA and IB in the presence of tris-(dipivalomethanato)europium $[Eu(DPM)_3]$. The addition of this reagent to the CCl_4 solution of IA or IB shifted the C_2 -methylene group into a spectral region where the spin-spin splitting of the C_2-H 's with the C_3-H becomes amenable to first-order analysis. (Figure 1b and 1Ib)^{11,***} According to Barfield and Grant, J_{gem} is dependent upon the dihedral angle between the methylene group and the π lobes of adjacent π bonds.¹² Assuming this type of relationship to exist in the present case, the J_{gem} of 17.3 Hz for the C_2 -methylene in IA can only be explained with a conformation in which the $C=O$ group bisects the C_2 -methylene group.⁺ This stereochemistry in combination with a planar lactone grouping necessitates that the cis isomer exist in the half-chair conformation IA. The vicinal coupling



IA



IB

$J_{2ax,3ax}=10.3$, $J_{2ax,3eq}=5.8$, and the long range coupling of 1.9 Hz observed between C_2 -Heq and C_4 -Heq are also in accord with this assignment. The large long range coupling between C_2 -Heq and C_4 -Heq is particularly revealing since the geometry in the half-chair conformation IA has these protons in the planar W configuration necessary for maximum effect.¹³ The geometry of a boat or half-boat conformation for the cis-isomer is not favorable for the observation of such a large long range $J_{2eq,4eq}$ coupling. In the case of the trans-isomer the slightly lower $J_{2eq,4eq}$ 16.1, the slightly larger J_{vic} (9.0 and 5.6 Hz) than expected and the absence of C_2-H_{eq} , C_4-H_{eq}

* Johnson, Starkovsky and Gurowitz have reported that in cyclohexanones, equatorial methyl groups appear at higher field and have coupling constants smaller than axial methyl groups.⁹

** Deshielding in the order of 0.18 ppm has been observed in cyclohexanols in going from 1,3-H,H to 1,3- CH_3 ,H interactions.¹⁰

*** Coupling constants are not significantly changed by the addition of $Eu(DPM)_3$ in the concentration range reported.

⁺ The sign of J_{gem} in these lactones is opposite the J_{vic} and is assumed to be negative.

long-range coupling is best accounted for if IB has a slightly flatten half-chair conformation similar to the structure proposed by Sheppard and Turner to explain the spectral properties of some steroidal lactones.² This flattening causes the C₃-CH₃ and the C₂-methylene protons to adopt a "pseudoaxial" and "pseudoequatorial" character in order to relieve the unfavorable steric interaction of the axial C₃-CH₃ group in IB. The vicinal, geminal and long range coupling of IA and IB as well as the observation that the NMR spectra (CDCl₃) are not significantly changed in the temperature range of -55 to 30° are consistent with a fixed conformation for these lactones, rather than a rapidly equilibrating set of conformers.* Both the cis and trans lactones I had infrared carbonyl absorption (CCl₄) at 1736 cm⁻¹ which is within the range (1730-1750 cm) suggested for the normal frequency for δ-lactones having a half-chair conformation.⁷

Acknowledgement. The authors are grateful to Dr. S. G. Levine, North Carolina State University at Raleigh, North Carolina, for many helpful suggestions and for a sample of Eu(DPM)₃. We also thank Dr. M. E. Wall, Director, of this laboratory for his support of this work.

*The J_{gem} and J_{vic} of IB might be accounted for by a rapidly equilibrating system. However, the fact that the C₅-CH₃ appears at δ1.33 in both IA and IB would be inconsistent with this interpretation.

References

1. This work was carried out under Contract PH43-65-1057 of the National Institutes of General Medical Sciences, National Institutes of Health, Bethesda, Maryland.
2. R. C. Sheppard and S. Turner, Chem. Comm., 77 (1968).
3. R. N. Johnson and N. V. Riggs, Tetrahedron Letters, 5119 (1967).
4. G. Di Maio, P. A. Tardella and C. Iavarone, Tetrahedron Letters, 2825 (1966).
5. J. F. McConnell, A. McL. Mathieson and B. P. Schoenborn, Tetrahedron Letters, 445 (1962).
6. A. McL. Mathieson and J. C. Taylor, Tetrahedron Letters, 590 (1961).
7. K. K. Cheung, K. H. Overton and G. A. Sim, Chem. Comm., 634 (1965).
8. H. Wolf, Tetrahedron Letters, 5151 (1966).
9. F. Johnson, N. A. Starkovsky and W. D. Gurowitz, J. Am. Chem. Soc., 87, 3492 (1965).
10. H. Booth, Tetrahedron, 22, 615 (1966).
11. J. K. M. Sanders and D. H. Williams, Chem. Comm., 422 (1970).
12. M. Barfield and D. M. Grant, J. Am. Chem. Soc., 85, 1899 (1963).
13. L. M. Jackman and S. Sternhell, Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry, 2nd ed., Pergamon Press, Inc., Elmsford, New York, page 334.